

**Q1.** Stars go through a life cycle.

Some stars will finish their life cycle as a black dwarf and other stars as a black hole.

(a) The table below gives the mass, relative to the Sun, of three stars, **J**, **K** and **L**.

Star	Mass of the star relative to the Sun
<b>J</b>	0.5
<b>K</b>	14.5
<b>L</b>	20.0

Which **one** of the stars, **J**, **K** or **L**, will become a black dwarf?

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Give a reason for your answer.

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(2)

(b) Scientists can take the measurements needed to calculate the mass of many stars.

Scientists cannot calculate the mass of the star Betelgeuse.

They estimate that the star has a mass between 8 and 20 times the mass of the Sun.

(i) Betelgeuse is in the red super giant stage of its life cycle.

What will happen to Betelgeuse at the end of the red super giant stage?

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(1)

(ii) Suggest **one** reason why scientists can only estimate and **not** calculate the mass of Betelgeuse.

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(1)

- (iii) In the future, it may become possible for scientists to calculate the mass of Betelgeuse.

Suggest **one** reason why.

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(1)

- (c) Describe what happens to a star, after the main sequence period, for the star to eventually become a **black dwarf**.

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(5)  
(Total 10 marks)

**Q2.** The table gives information about the three types of particle that make up an atom.

Particle	Relative mass	Relative charge
Proton		+1
Neutron	1	
Electron	very small	-1

(a) Complete the table by adding the **two** missing values.

(2)

(b) Use the information in the table to explain why an atom has no overall electrical charge.

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(2)

(c) Uranium has two natural isotopes, uranium-235 and uranium-238. Uranium-235 is used as a fuel inside a nuclear reactor. Inside the reactor, atoms of uranium-235 are split and energy is released.

(i) How is the structure of an atom of uranium-235 different from the structure of an atom of uranium-238?

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(1)

(ii) The nucleus of a uranium-235 atom must absorb a particle before the atom is able to split.

What type of particle is absorbed?

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(1)

(iii) The nucleus of an atom splits into smaller parts in a reactor.

What name is given to this process?

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(1)  
(Total 7 marks)

**Q3.** *To gain full marks in this question you should write your ideas in good English. Put them into a sensible order and use the correct scientific words.*

Explain briefly how stars like the Sun are thought to have been formed.

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(Total 2 marks)

**Q4.** The process of nuclear fusion results in the release of energy.

(a) (i) Describe the process of nuclear fusion.

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(2)

(ii) Where does nuclear fusion happen naturally?

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(1)

(b) For many years, scientists have tried to produce a controlled nuclear fusion reaction that lasts long enough to be useful. However, the experimental fusion reactors use more energy than they produce.

(i) From the information given, suggest **one** reason why nuclear fusion reactors are not used to produce energy in a nuclear power station.

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(1)

(ii) Suggest **one** reason why scientists continue to try to develop a practical nuclear fusion reactor.

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(1)

(Total 5 marks)

**Q5.(a)** Nuclear fission is used in nuclear power stations to generate electricity. Nuclear fusion happens naturally in stars.

(i) Explain briefly the difference between *nuclear fission* and *nuclear fusion*.

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**(2)**

(ii) What is released during both nuclear fission and nuclear fusion?

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**(1)**

(b) Plutonium-239 is used as a fuel in some nuclear reactors.

(i) Name another substance used as a fuel in some nuclear reactors.

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**(1)**

(ii) There are many isotopes of plutonium.

What do the nuclei of different plutonium isotopes have in common?

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**(1)**

**(Total 5 marks)**

**Q6.** Nuclear fission and nuclear fusion are two processes that release energy.

- (a) (i) Use the correct answer from the box to complete each sentence.

**Geiger counter      nuclear reactor      star**

Nuclear fission takes place within a .....

Nuclear fusion takes place within a .....

(2)

- (ii) State **one** way in which the process of nuclear fusion differs from the process of nuclear fission.

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(1)

- (b) The following nuclear equation represents the fission of uranium-235 (U-235).



Chemical symbols:

Ba - barium

Kr - krypton

- (i) Use the information in the equation to describe the process of nuclear fission.

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(4)

- (ii) An isotope of barium is Ba-139.  
Ba-139 decays by beta decay to lanthanum-139 (La-139).

Complete the nuclear equation that represents the decay of Ba-139 to La-139.



(3)

(Total 10 marks)



**Q7.** Stars go through a life cycle. About 90 % of all stars are in the 'main sequence' period of the life cycle.

- (a) Stars are stable during the 'main sequence' period of the life cycle.

Why?

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(1)

- (b) The table gives an estimated time for the number of years that three stars, **X**, **Y** and **Z**, will be in the 'main sequence' period of their life cycle.

Star	Relative mass of the star compared to the Sun	Estimated 'main sequence' period in millions of years
<b>X</b>	0.1	4 000 000
<b>Y</b>	1.0	9 000
<b>Z</b>	40.0	200

- (i) This data suggests that there is a pattern linking the mass of a star and the number of years the star is in the 'main sequence' period of its life cycle.

What is the pattern suggested by the data?

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(1)

- (ii) Scientists cannot give the exact number of years a star will be in the 'main sequence' period.

Suggest why.

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(1)

- (iii) Nuclear fusion is the process by which energy is released in stars.

Which **one** of the following can be concluded from the data in the table?

Draw a ring around the correct answer in the box to complete the sentence.

The rate of nuclear fusion in a large star is 

faster than
the same as
slower than

 in a small star.

Explain the reason for your answer.

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(3)

- (c) *In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate.*

Describe what happens to a star **much bigger** than the Sun, once the star reaches the end of the 'main sequence' period of its life cycle.

Your answer should include the names of the stages the star passes through.

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(6)

(Total 12 marks)